

Synthesis of Analogues of Brassinosteroids from Chenodeoxycholic Acid

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The synthesis and spectroscopic characterization of two new bioactive analogues of brassinosteroids with a 24-hydroxylated cholanolic side chain, an A/B ring *cis*-junction and oxygenated functions in C-7 is described.

Introduction

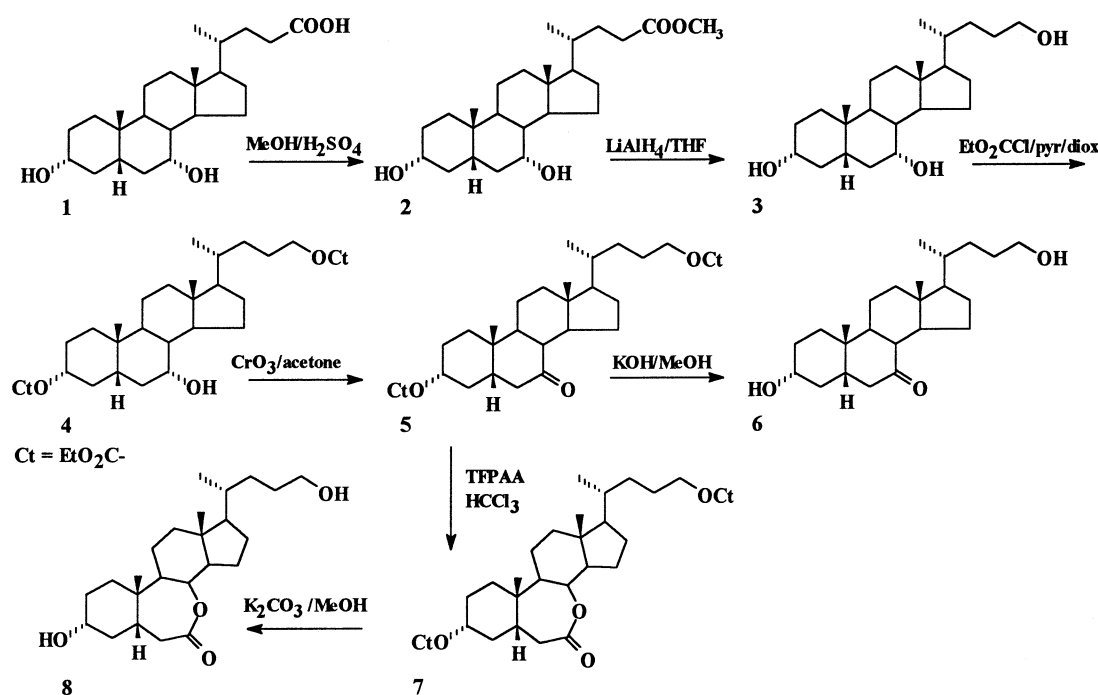
Brassinosteroids are plant hormones^[1] with remarkable biological activity that have stimulated many laboratories to undertake their synthesis^[2] and the study of the relationship between molecular structure and biological activity.^[3] Although it is generally accepted that a high brassinosteroid activity requires (a) a 2 α ,3 α -diol in the A ring, (b) a 6-one or 6-oxo-7-oxalactone moiety in the B ring, (c) an A/B *trans*-fused ring junction, (d) a (22*R*,23*R*)-diol and (e) a 24-methyl or -ethyl substituent,^[1] a great variety of bioactive analogues with modifications in the steroidal nucleus and side chain have been synthesized.^[4] Recently, we have reported on the synthesis of some bioactive spirostanoic^[5] and

furostanoic^[6] analogues of brassinosteroids starting from steroidal sapogenins. We now wish to describe the synthesis of two new bioactive analogues of brassinosteroids with a 24-hydroxylated cholanolic side chain, an A/B ring *cis*-junction and oxygenated functions in C-7.

Results and Discussion

Methyl chenodeoxycholate (**2**) was obtained by a procedure already described.^[7] LiAlH₄ reduction of **2** in tetrahydrofuran (THF) led to the triol **3**^[8] which was selectively acetylated to afford the 7 α -hydroxy-5 β -cholan-3 α ,24-diyl dicathylate (**4**). Oxidation of **4**, followed by hydrolysis with 5% methanolic KOH afforded 3 α ,24-dihydroxy-5 β -cholan-

Scheme 1. Synthetic sequence



7-one (**6**) in 79% overall yield. Baeyer-Villiger reaction of **5** with trifluoroperoxyacetic acid (TFPAA) obtained "in situ" by the trifluoroacetic anhydride (TFAA)/hydrogen peroxide procedure,^[8] led to pure lactone **7**. Selective hydrolysis of **7** with 1% methanolic K₂CO₃ during 15 hours afforded the desired compound 3 α ,24-dihydroxy-B-homo-8-oxa-5 β -cholan-7-one (**8**) in 66% overall yield (Scheme 1).

The biological activity of the brassinosteroid analogues **6** and **8** was examined by the Radish test.^[9] Compound **6** produced, at the concentration of 10⁻⁶ μ g/ml, an increase of 36% (referred to an untreated control) in the weight of cotyledons and at 10⁻⁴ and 10⁻⁶ μ g/ml an increase of 16 and 17% respectively, in the length of hypocotyls. Compound **8** produced at the concentration of 10⁻⁴ μ g/ml just a small increase of 9% in the weight of cotyledons.

Conclusions

Two new analogues of brassinosteroids **6** and **8** with an A/B ring *cis* junction, oxygenated functions in C-7 and a 24-hydroxylated cholan side chain were synthesized from chenodeoxycholic acid (**1**). In contrast with the expected results, the compounds which bear drastic structural modifications, showed biological activity. The biological activity of the 7-oxo compound **6** proved to be higher than that of the lactone **8**.

Experimental Section

General: IR spectra were recorded from KBr cells with a Phillips Analytical PU9800 FT-IR spectrometer. – NMR: Bruker 250 ACF (250 MHz and 62.9 MHz, for ¹H and ¹³C, respectively). For ¹H NMR, CDCl₃ or CD₃OD (solvents) and TMS (internal standard); for ¹³C NMR, CDCl₃ or CD₃OD (δ_C = 77.0 and 49.0) were used. Temperature of the samples was 300 K and concentrations were 30 mg/ml for ¹³C spectra and 10 mg/ml for ¹H spectra. – MS: JEOL JMS-DX 300 (70 eV). – Melting points were measured with Electrothermal 9100 equipment and are uncorrected.

5 β -Cholane-3 α ,7 α ,24-triol (3**):** To a suspension of 500 mg of LiAlH₄ in 15 ml of THF, a solution of 2.54 g (6.72·10⁻³ mol) of methyl chenodeoxycholanate (**2**) in 15 ml of THF was added dropwise, and the mixture was stirred at reflux temp. for 4 h, poured into cool water and 10% H₂SO₄ (10 ml) was added. The product was extracted with ethyl acetate (3 \times 30 ml) and the combined organic extracts were washed with brine (2 \times 20 ml) and water (2 \times 20 ml), dried with MgSO₄ and concentrated to give 2.25 g of **3** (95%), colorless crystals, m.p. 104.8–105.8°C. – IR (KBr): $\tilde{\nu}$ = 3404 cm⁻¹ (OH). – ¹H NMR (CD₃OD): δ = 3.51 (m, 1 H, 3-H_{ax}), 3.79 (m, 1 H, 7-H_{eq}), 0.70 (s, 3 H, 18-H), 0.93 (s, 3 H, 19-H), 0.97 (d, *J* = 6.21 Hz, 3 H, 21-H), 3.40 (t, 2 H, 24-H). – ¹³C NMR (CD₃OD): δ = 36.9 (C-1), 33.6 (C-2), 73.2 (C-3), 41.5 (C-4), 43.6 (C-5), 36.3 (C-6), 69.4 (C-7), 41.2 (C-8), 34.4 (C-9), 36.6 (C-10), 22.2 (C-11), 40.8 (C-12), 44.0 (C-13), 51.9 (C-14), 25.1 (C-15), 29.8 (C-16), 57.9 (C-17), 12.7 (C-18), 23.9 (C-19), 37.5 (C-20), 19.7 (C-21), 31.8 (C-22), 30.7 (C-23), 63.9 (C-24). – MS (70 eV); *m/z* (%): 378 (4.9) [M⁺], 360 (33.6) [M⁺ – H₂O], 342 (71.3) [M⁺ – 2H₂O], 283 (2.8) [M⁺ – 2 H₂O – C₃H₇O], 255 (28.7) [M⁺ – 2 H₂O – C₅H₁₁O], 41 (100) [C=C–C⁺]. – C₂₄H₄₂O₃ (378.59): calcd. C 76.14, H 11.18; found C 76.10, H 11.24.

7 α -Hydroxy-5 β -cholane-3 α ,24-diyl Dicathylate (4**):** To a solution of 1.65 g (4.36·10⁻³ mol) of **3** in 15 ml of dioxane and 8 ml of

pyridine at 0°C, 10 ml of ethyl chloroformate was added dropwise, and the mixture was stirred at room temp. for 2 h. 8 ml of 32% HCl and 50 ml of water were added and the product was extracted with dichloromethane. The organic phase was washed with water, dried with MgSO₄ and the solvent removed under reduced pressure to give 2.21 g of **4** (97%), white crystalline solid, m.p. 107.4–109.4°C. – IR (KBr): $\tilde{\nu}$ = 3536 cm⁻¹ (OH), 1740 (C=O). – ¹H NMR (CDCl₃): δ = 4.44 (m, 1 H, 3-H_{ax}), 3.84 (m, 1 H, 7-H_{eq}), 0.66 (s, 3 H, 18-H), 0.92 (s, 3 H, 19-H), 0.93 (d, *J* = 7.0 Hz, 3 H, 21-H), 4.09 (t, 2 H, 24-H), 1.29–1.31 (t, 6 H, *J* = 7.1 Hz, CH₃CH₂OCO₂), 4.16–4.19 (q, *J* = 7.2 Hz and *J* = 7.1 Hz, 4 H, CH₃CH₂OCO₂). – ¹³C NMR (CDCl₃): δ = 34.9 (C-1), 26.6 (C-2), 78.1 (C-3), 35.1 (C-4), 41.1 (C-5), 34.3 (C-6), 68.3 (C-7), 39.4 (C-8), 32.7 (C-9), 35.0 (C-10), 20.5 (C-11), 39.6 (C-12), 42.6 (C-13), 50.3 (C-14), 23.6 (C-15), 28.2 (C-16), 55.8 (C-17), 11.7 (C-18), 22.6 (C-19), 35.4 (C-20), 18.4 (C-21), 31.7 (C-22), 25.2 (C-23), 68.4 (C-24), 14.3 double intensity (CH₃–CH₂–O), 63.5/63.8 (CH₃–CH₂–O), 155.3/154.7 (O₂C=O).

3 α ,24-Dicathylxy-5 β -cholane-7-one (5**):** To a solution of 2 g of **4** (3.85·10⁻³ mol) in 15 ml of acetone at 0°C, 1.5 ml of Jones reagent was added dropwise and the mixture was stirred for 10 min. 0.5 ml of 2-propanol and enough water to produce a precipitate was added. The precipitate was collected by filtration, washed with abundant water and dried at 60°C to give 1.85 g of **5** (98%), syrup. – IR (KBr): $\tilde{\nu}$ = 1744, 1715 cm⁻¹ (C=O). – ¹H NMR (CDCl₃): δ = 4.54 (m, 1 H, 3-H_{ax}), 0.65 (s, 3 H, 18-H), 1.21 (s, 3 H, 19-H), 0.93 (d, *J* = 6.4 Hz, 3 H, 21-H), 4.09 (t, 2 H, 24-H), 1.29–1.31 (t, *J* = 7.1 Hz, 6 H, CH₃CH₂OCO₂), 4.16–4.19 (q, *J* = 7.1 Hz, 4 H, CH₃CH₂OCO₂). – ¹³C NMR (CDCl₃): δ = 33.7 (C-1), 26.0 (C-2), 76.5 (C-3), 33.0 (C-4), 42.6 (C-5), 45.2 (C-6), 211.5 (C-7), 48.9 (C-8), 45.7 (C-9), 35.1 (C-10), 21.7 (C-11), 38.9 (C-12), 42.6 (C-13), 49.5 (C-14), 25.4 (C-15), 28.3 (C-16), 54.8 (C-17), 12.0 (C-18), 22.9 (C-19), 35.3 (C-20), 18.6 (C-21), 31.7 (C-22), 24.8 (C-23), 68.4 (C-24), 14.2/14.3 (CH₃–CH₂–O), 63.7/63.8 (CH₃–CH₂–O), 155.3/154.5 (O₂C=O).

3 α ,24-Dihydroxy-5 β -cholane-7-one (6**):** A mixture of 600 mg of **5** (1.59·10⁻³ mol) and 10 ml of 5% methanolic KOH was stirred at room temp. for 15 min and then poured into water. The solid was collected by filtration, washed with abundant water and dried at 60°C. Chromatographic separation on silica gel with hexane/ethyl acetate (3:2) as solvent system gave 391 mg of **6** (90%), white crystalline solid, m.p. 148–149°C. – IR (KBr): $\tilde{\nu}$ = 3496, 3350 cm⁻¹ (OH), 1713 (C=O). – ¹H NMR (CDCl₃): δ = 3.58 (m, 1 H, 3-H_{ax}), 0.66 (s, 3 H, 18-H), 1.20 (s, 3 H, 19-H), 0.94 (d, *J* = 6.5 Hz, 3 H, 21-H), 3.58 (t, 2 H, 24-H). – ¹³C NMR (CDCl₃): δ = 34.1 (C-1), 29.6 (C-2), 70.5 (C-3), 37.1 (C-4), 42.8 (C-5), 45.4 (C-6), 213.0 (C-7), 49.5 (C-8), 46.1 (C-9), 35.1 (C-10), 21.6 (C-11), 38.9 (C-12), 42.5 (C-13), 48.8 (C-14), 24.7 (C-15), 28.3 (C-16), 54.9 (C-17), 11.9 (C-18), 23.0 (C-19), 35.4 (C-20), 18.6 (C-21), 31.8 (C-22), 29.2 (C-23), 63.1 (C-24). – MS (70 eV); *m/z* (%): 376 (9.2) [M⁺], 358 (13.7) [M⁺ – H₂O], 340 (7.8) [M⁺ – 2H₂O], 299 (3.9) [M⁺ – H₂O – C₃H₇O], 271 (7.8) [M⁺ – H₂O – C₅H₁₁O], 41 (100) [C=C–C⁺]. – C₂₄H₄₀O₃ (376.58): calcd. C 76.55, H 10.71; found C 76.49, H 10.70.

3 α ,24-Dicathylxy-B-homo-8-oxa-5 β -cholane-7-one (7**):** To a solution of trifluoroperoxyacetic acid [prepared from 30% aqueous H₂O₂ (0.7 ml, 5.85 mmol) and TFAA (3.7 ml)] in 5 ml of chloroform, a solution of 216 mg of **6** (0.42·10⁻³ mol) in 15 ml of chloroform was added dropwise at 0°C and the mixture was stirred for 5 min and diluted with 10 ml of chloroform. The resulting solution was washed with water (1 \times 3 ml), aqueous Na₂CO₃ (1 \times 3 ml), aqueous NaHSO₃ (1 \times 3 ml), brine (1 \times 3 ml) and water (1 \times 3 ml),

and dried with MgSO_4 . The solvent was removed by evaporation to give 209 mg of **5** (94%), white crystalline solid, m.p. 92.4–94.4°C. – IR (KBr): $\tilde{\nu}$ = 1743 cm^{-1} (C=O). – ^1H NMR (CDCl_3): δ = 4.59 (m, 1 H, 3- H_{ax}), 4.18 (t, 1 H, 8- H_{ax}), 0.66 (s, 3 H, 18-H), 1.08 (s, 3 H, 19-H), 0.92 (d, J = 6.5 Hz, 3 H, 21-H), 4.09 (t, 2 H, 24-H), 1.29–1.31 (t, J = 7.1 Hz, 6 H, $\text{CH}_3\text{CH}_2\text{OCO}_2$), 4.18 (q, J = 7.1 Hz, 4 H, $\text{CH}_3\text{CH}_2\text{OCO}_2$). – ^{13}C NMR (CDCl_3): δ = 35.3 (C-1), 26.4 (C-2), 76.3 (C-3), 31.6 (C-4), 41.5 (C-5); 35.5 (C-6), 173.9 (C-7), 79.6 (C-8), 39.9 (C-9), 37.5 (C-10), 22.5 (C-11), 38.4 (C-12), 43.2 (C-13), 54.4 (C-14), 25.1 (C-15), 27.8 (C-16), 55.8 (C-17), 11.5 (C-18), 23.4 (C-19), 35.1 (C-20), 18.4 (C-21), 31.7 (C-22), 25.3 (C-23), 68.3 (C-24), 14.2/14.3 ($\text{CH}_3\text{--CH}_2\text{--O}$), 63.8 ($\text{CH}_3\text{--CH}_2\text{--O}$), 155.3/154.5 ($\text{O}_2\text{C=O}$).

3 α ,24-Dihydroxy-B-homo-8-oxa-5 β -cholan-7-one (8): A mixture of 200 mg of **7** ($0.51 \cdot 10^{-3}$ mol) and 10 ml of 1% methanolic K_2CO_3 was stirred at room temp. for 15 h and then poured into water. The solid was collected by filtration, washed with abundant water and dried at 60°C. Chromatographic separation on silica gel with hexane/ethyl acetate (3:2) as solvent system gave 130 mg of **8** (89%), white crystalline solid, m.p. 119.0–119.4°C. – IR (KBr): $\tilde{\nu}$ = 3342 cm^{-1} (OH), 1743 (C=O). – ^1H NMR (CDCl_3): δ = 3.58 (m, 1 H, 3- H_{ax}), 4.21 (t, 1 H, 8- H_{ax}), 0.67 (s, 3 H, 18-H), 1.07 (s, 3 H, 19-H), 0.93 (d, J = 6.4 Hz, 3 H, 21-H), 3.58 (t, 2 H, 24-H). – ^{13}C NMR (CDCl_3): δ = 35.6 (C-1), 29.9 (C-2), 70.4 (C-3), 35.7 (C-4), 41.6 (C-5) 35.6 (C-6) 174.9 (C-7), 80.0 (C-8), 39.8 (C-9), 37.4 (C-10), 22.4 (C-11), 38.4 (C-12), 43.1 (C-13), 54.3 (C-14), 25.0 (C-15), 27.8 (C-16), 55.9 (C-17) 11.4 (C-18), 23.4 (C-19), 35.2 (C-20), 18.5 (C-21), 31.7 (C-22), 29.3 (C-23), 63.1 (C-24). – MS (70 eV); m/z

(%): 392 (2.9) [M^+], 374 (8) [$\text{M}^+ - \text{H}_2\text{O}$], 356 (7.2) [$\text{M}^+ - 2\text{H}_2\text{O}$], 315 (1.5) [$\text{M}^+ - \text{H}_2\text{O} - \text{C}_3\text{H}_7\text{O}$], 305 (4.3) [$\text{M}^+ - \text{C}_5\text{H}_{11}\text{O}$], 287 (4.3) [$\text{M}^+ - \text{H}_2\text{O} - \text{C}_5\text{H}_{11}\text{O}$], 41 (100) [$\text{C}=\text{C}-\text{C}^+$]. – $\text{C}_{24}\text{H}_{40}\text{O}_4$ (392.58): calcd. C 73.43, H 10.27; found C 73.49, H 10.35.

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